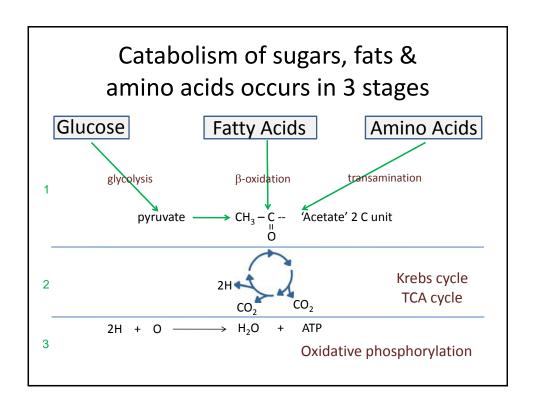
Oxidative Phosphorylation

(Cytochrome Chain)

Barbara Moreland





Examples of oxidation reactions

1. Add O₂ directly to the molecule (rare event in metabolism)

Oxidation example 1

Direct addition of O₂

xanthine uric acid

Examples of oxidation reactions

- 1. Add O₂ directly to the molecule (rare event in metabolism)
- 2. Remove one electron e⁻

Oxidation example 2

Removal of one electron

$$Fe^{2+} \xrightarrow{-e^{-}} Fe^{3+}$$

Reduced form Oxidised form (bright red) (brown)

Iron in haemoglobin or cytochrome proteins

Examples of oxidation reactions

- 1. Add O₂ directly to the molecule (rare event in metabolism)
- 2. Remove one electron e
- 3. Remove a pair of hydrogen atoms

Oxidation example 3

Hydrogen transfer

Ethanol + X ==== Ethanol + XH₂

Reduced Oxidised Oxidised Reduced form cofactor product cofactor

Cofactor required as hydrogen acceptor

Examples of hydrogen acceptor cofactors

NAD⁺ FAD NADP⁺